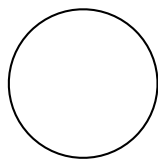




$$\varphi_1 = \varphi_{sA} + \varphi_{sB}$$

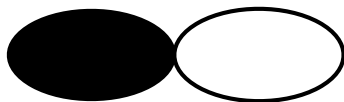


$$\varphi_2 = \varphi_{sA} - \varphi_{sB}$$



$$\varphi_3 \equiv 2s$$

Be



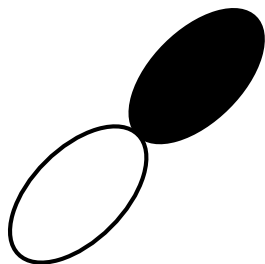
$$\varphi_4 \equiv 2p_z$$

Be



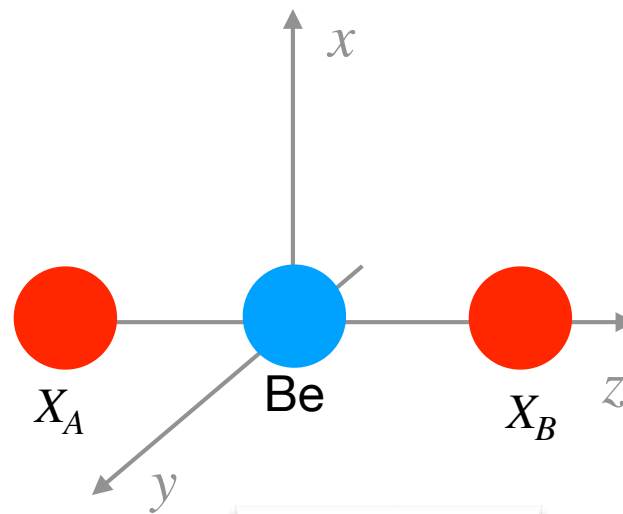
$$\varphi_5 \equiv 2p_x$$

Be



$$\varphi_6 \equiv 2p_y$$

Be

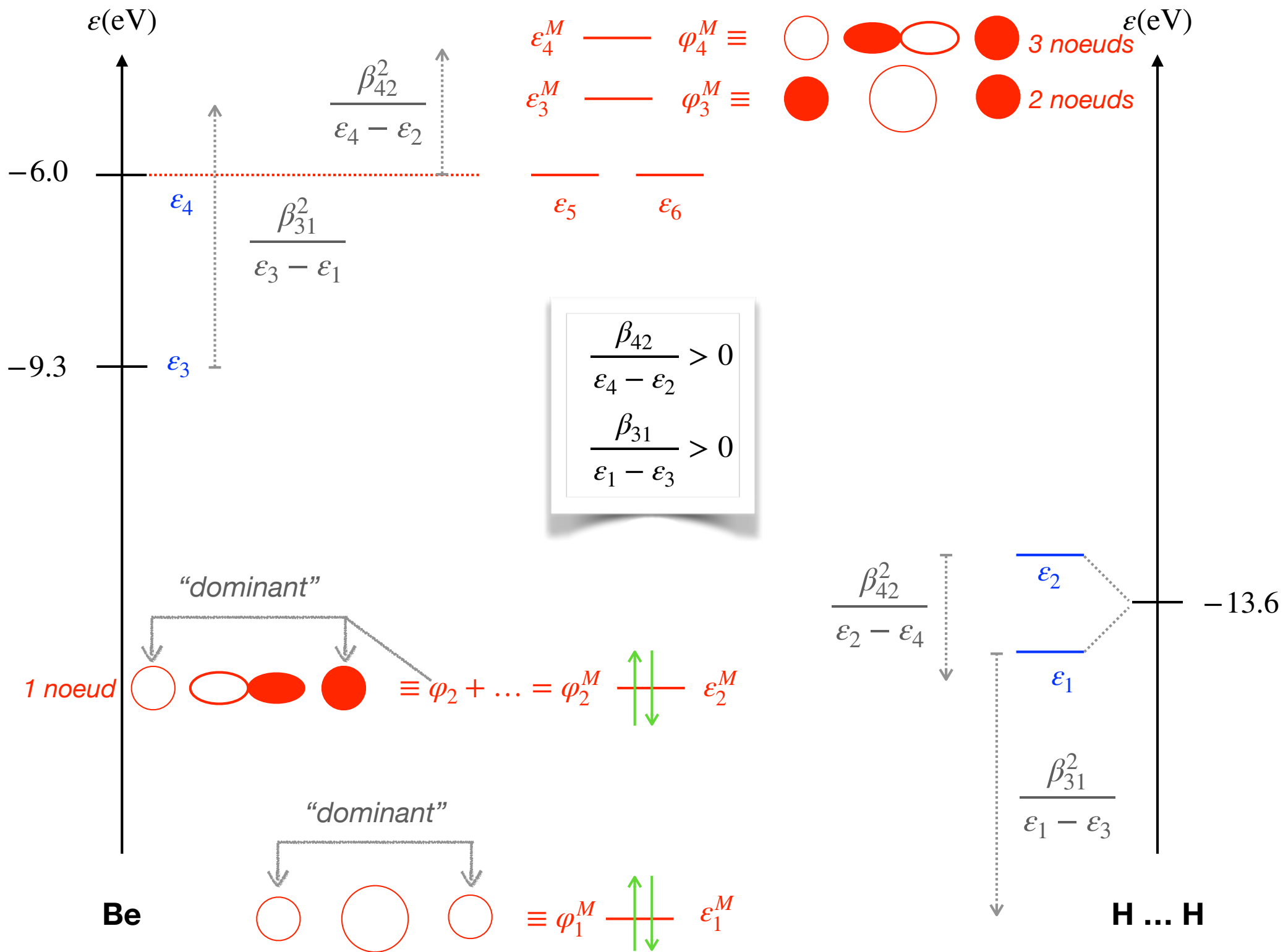


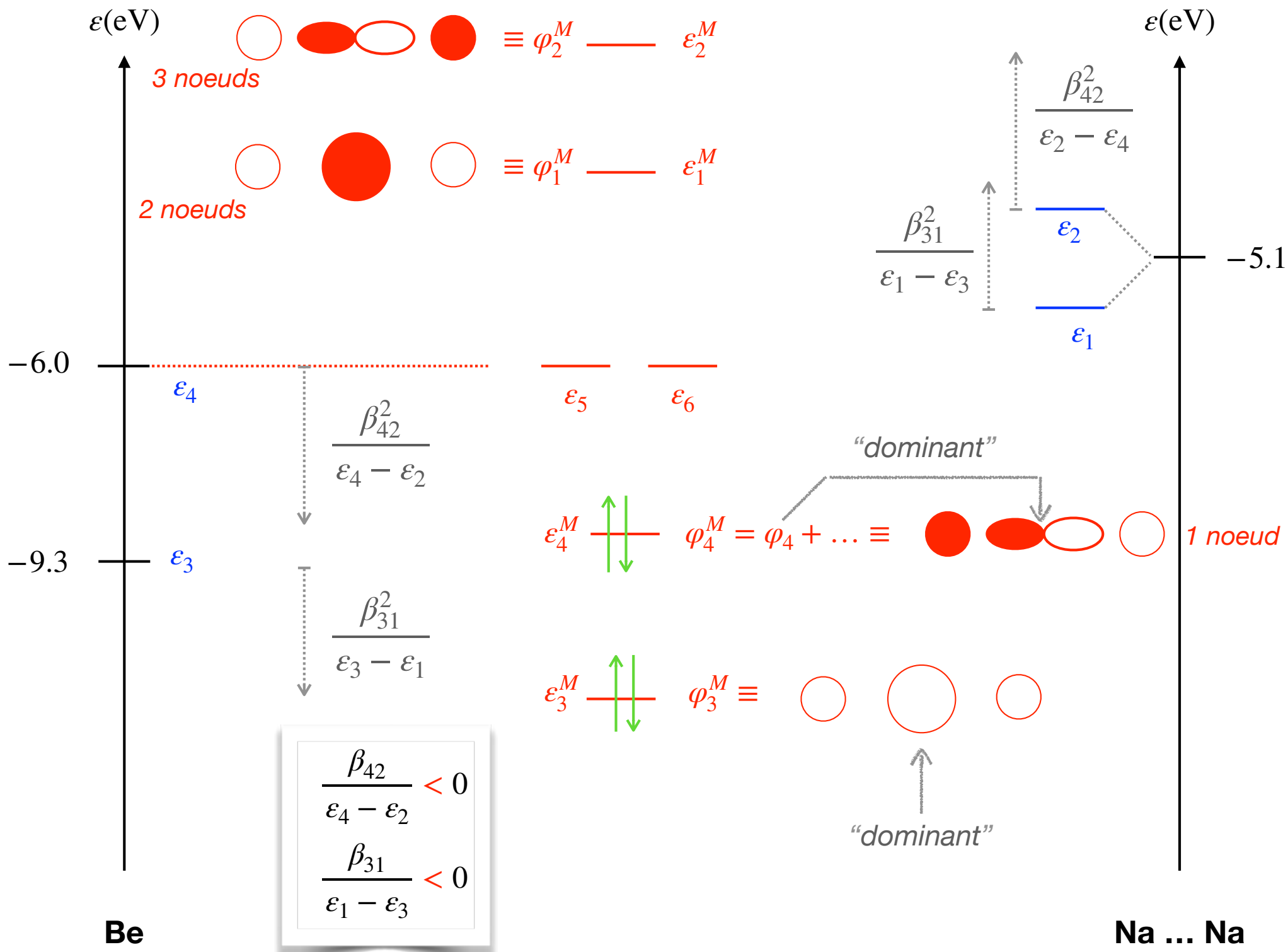
X = H, Na

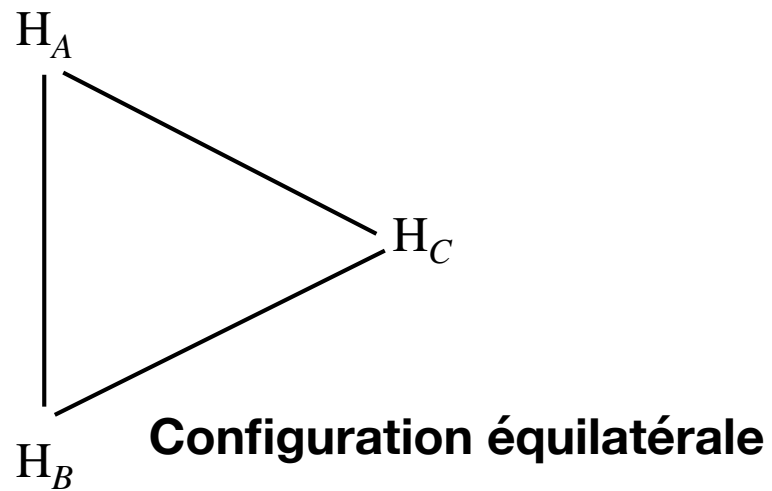
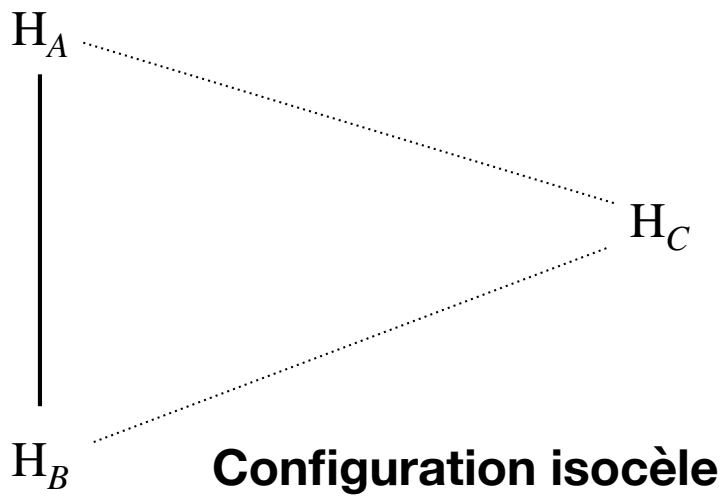
$$\int d\mathbf{r} \varphi_3(\mathbf{r}) \times \varphi_{s_A}(\mathbf{r}) \equiv \int d\mathbf{r} \begin{array}{c} \text{○} \\ X_A \end{array} \times \begin{array}{c} \text{○} \\ \text{Be} \end{array} = \int d\mathbf{r} \begin{array}{c} \text{○} \\ \text{Be} \end{array} \times \begin{array}{c} \text{○} \\ X_B \end{array} = \int d\mathbf{r} \varphi_3(\mathbf{r}) \times \varphi_{s_B}(\mathbf{r}) > 0$$

$$\int d\mathbf{r} \varphi_4(\mathbf{r}) \times \varphi_{s_A}(\mathbf{r}) \equiv \int d\mathbf{r} \begin{array}{c} \text{○} \\ X_A \end{array} \times \begin{array}{c} \text{●} \quad \text{○} \\ \text{Be} \end{array} = - \int d\mathbf{r} \begin{array}{c} \text{●} \quad \text{○} \\ \text{Be} \end{array} \times \begin{array}{c} \text{○} \\ X_B \end{array}$$

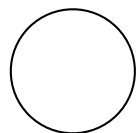
$$= - \int d\mathbf{r} \varphi_4(\mathbf{r}) \times \varphi_{s_B}(\mathbf{r}) < 0$$





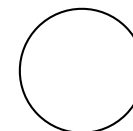


$$\varphi_1 = \varphi_{sA} + \varphi_{sB} \equiv$$

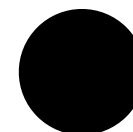
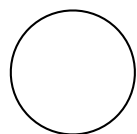


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$$\varphi_2 = \varphi_{sA} - \varphi_{sB} \equiv$$

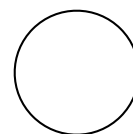


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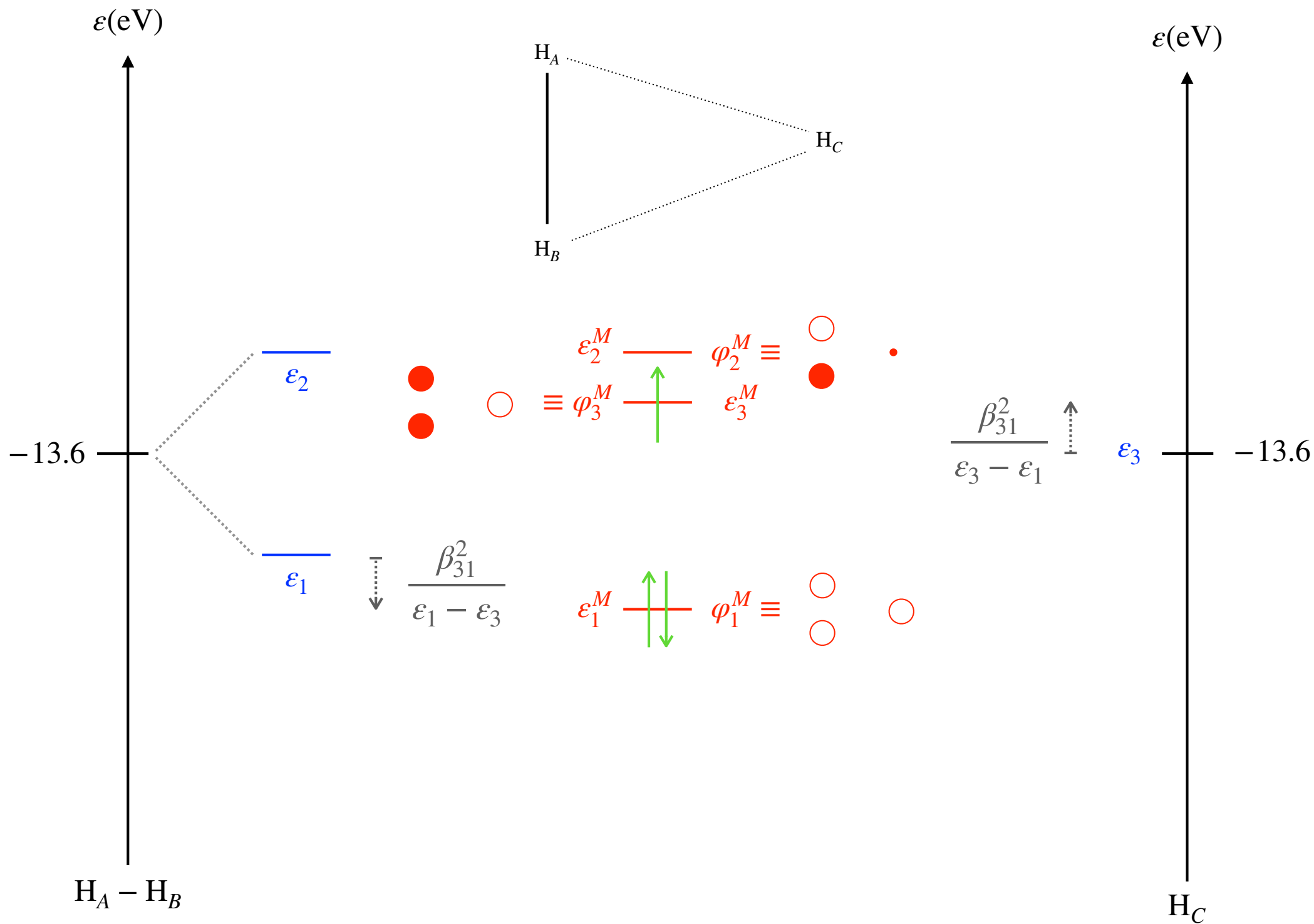


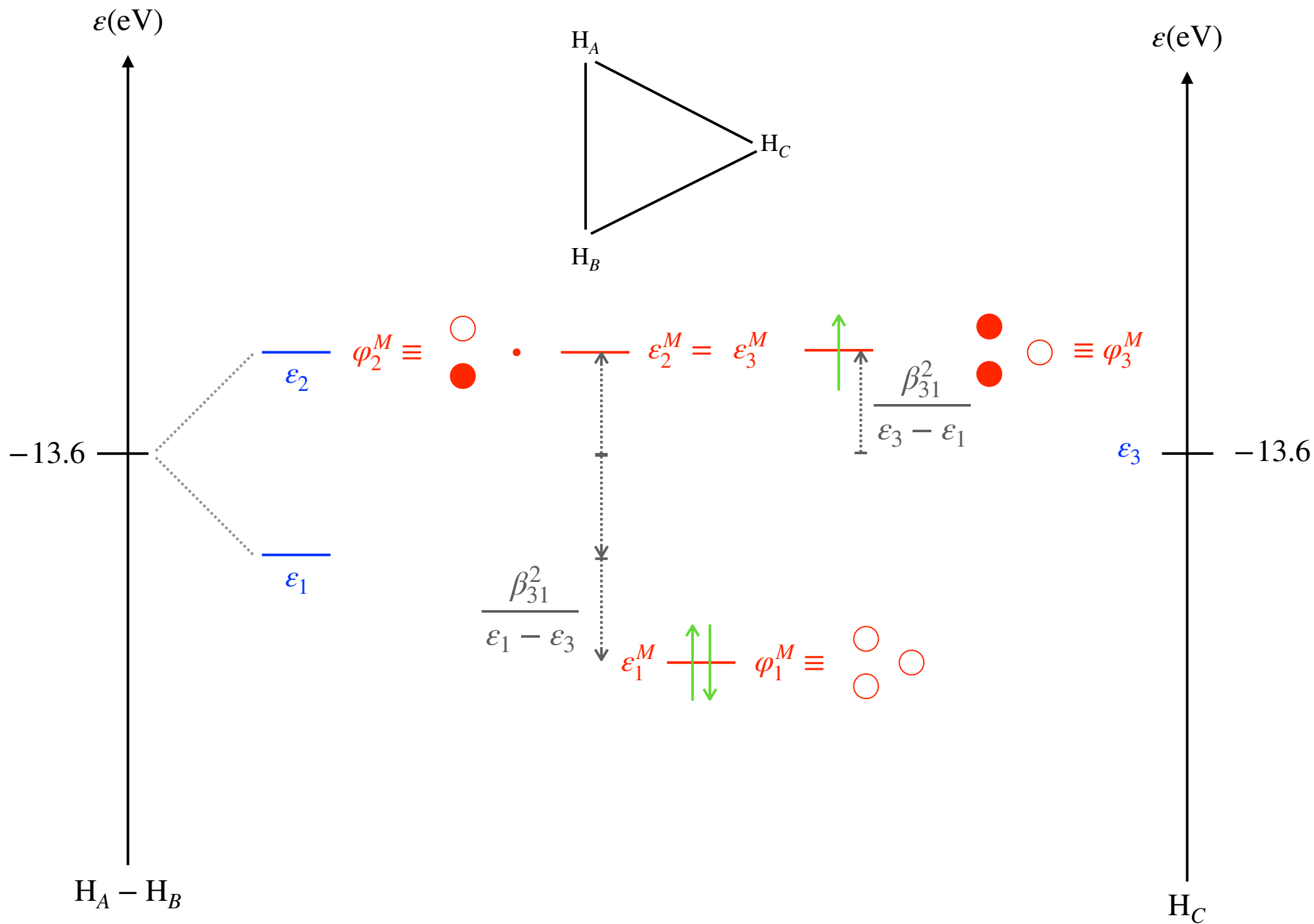
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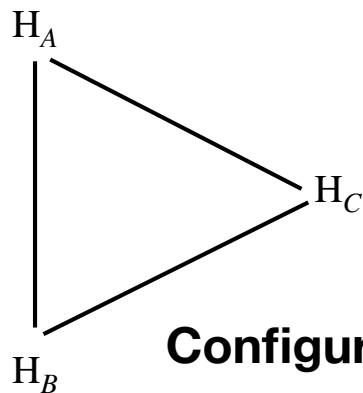
$$\varphi_3 = \varphi_{sC} \equiv$$



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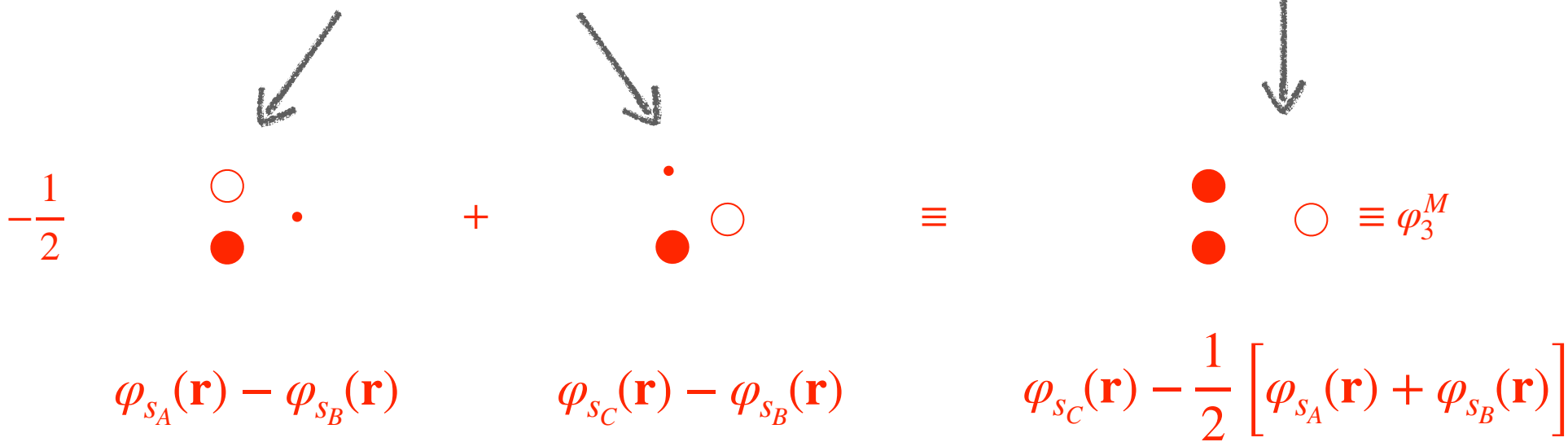


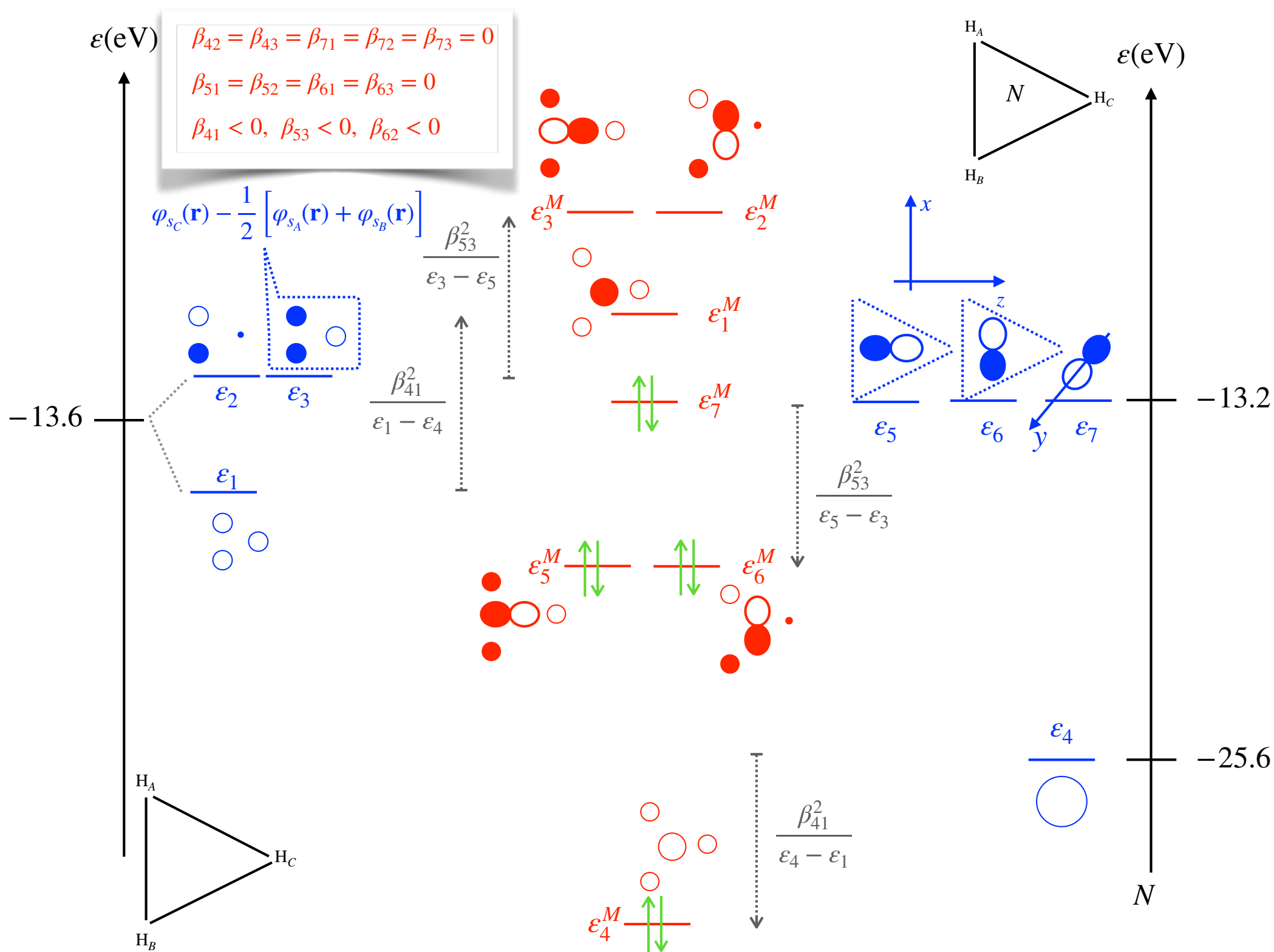


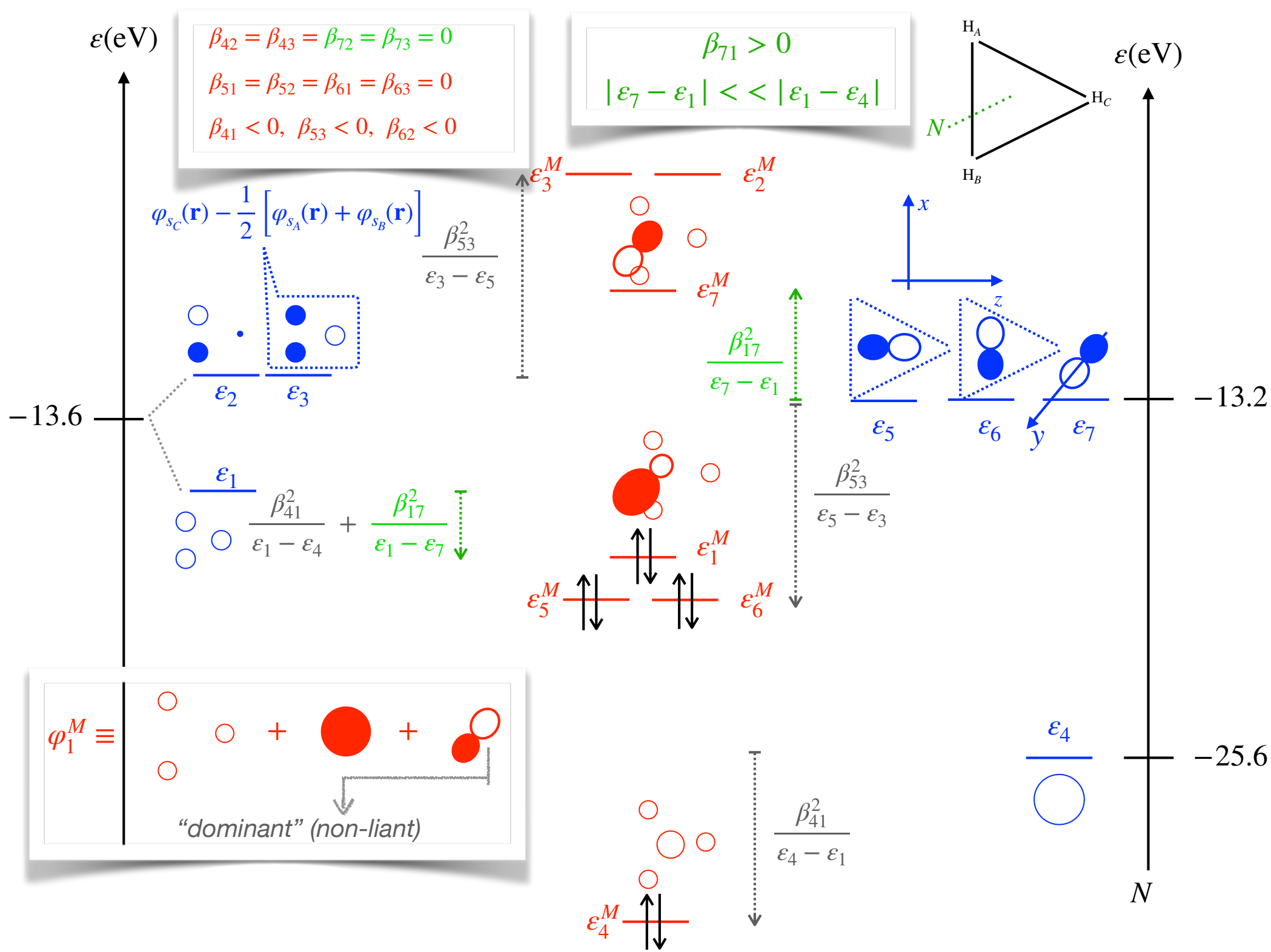


Orbitales “équivalentes” (et donc combinables) d'énergie ε_2^M

Orbitale de même énergie $\varepsilon_3^M = \varepsilon_2^M$



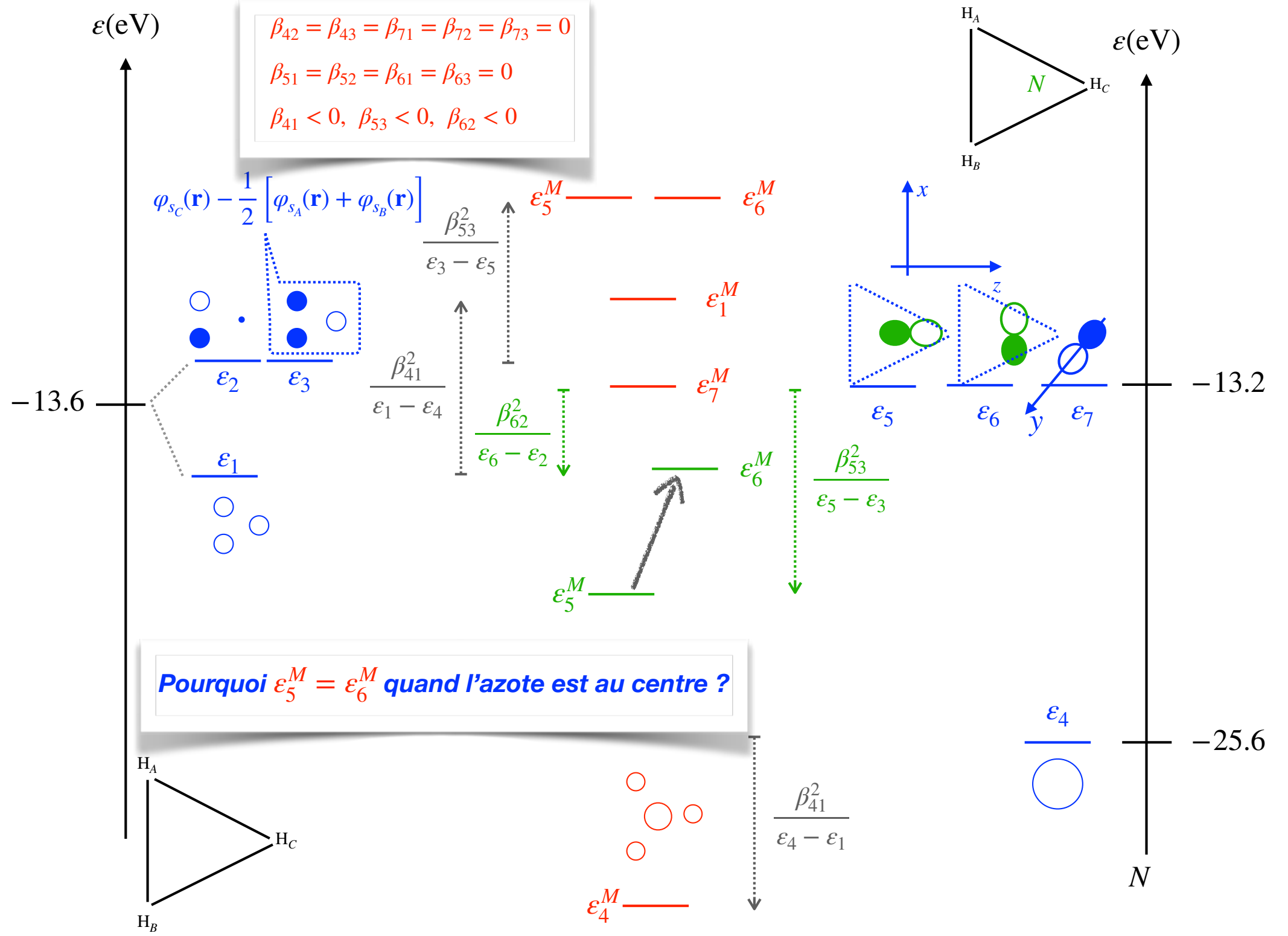




$$\beta_{42} = \beta_{43} = \beta_{71} = \beta_{72} = \beta_{73} = 0$$

$$\beta_{51} = \beta_{52} = \beta_{61} = \beta_{63} = 0$$

$$\beta_{41} < 0, \beta_{53} < 0, \beta_{62} < 0$$

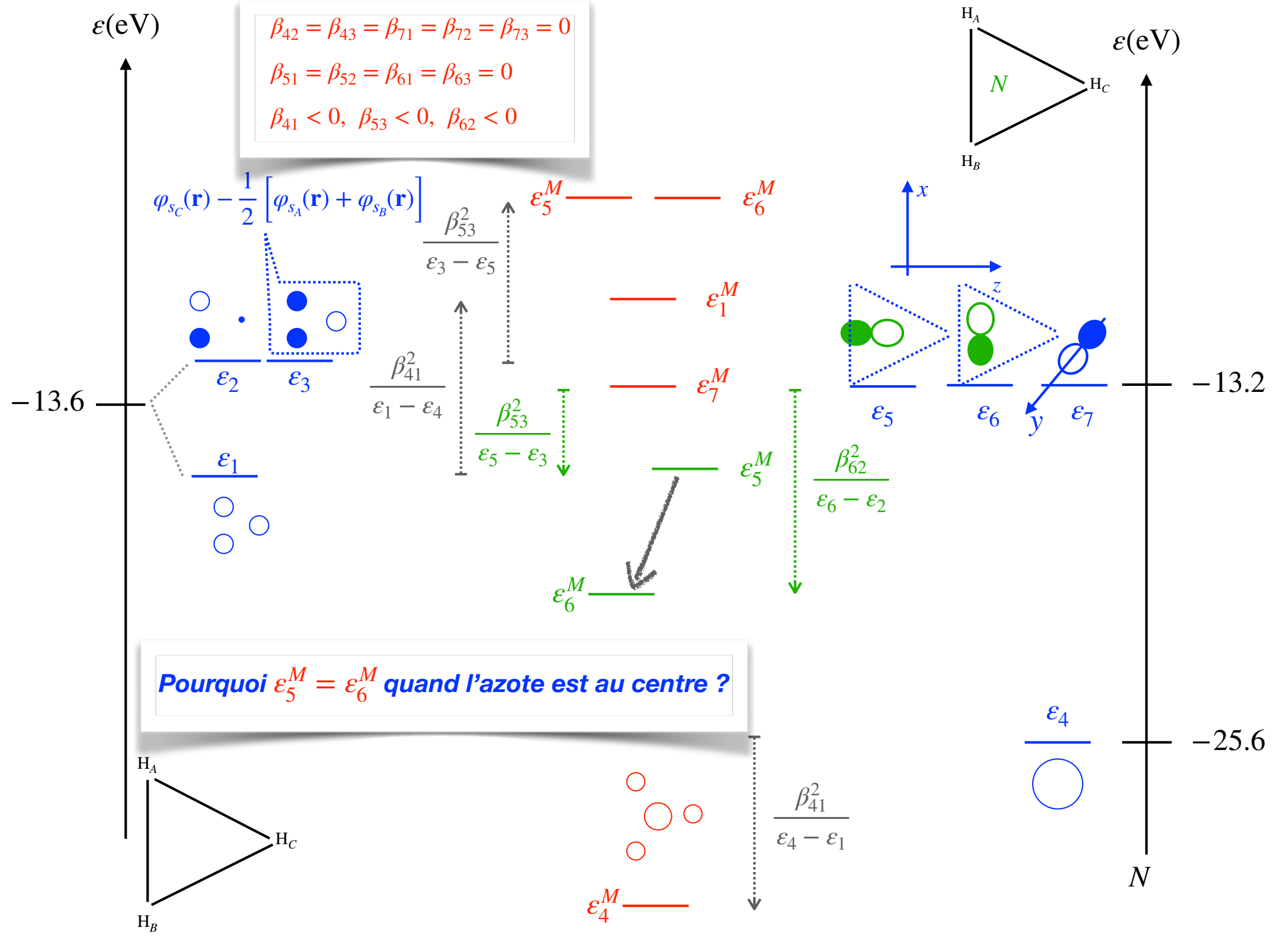


Pourquoi $\epsilon_5^M = \epsilon_6^M$ quand l'azote est au centre ?

$$\beta_{42} = \beta_{43} = \beta_{71} = \beta_{72} = \beta_{73} = 0$$

$$\beta_{51} = \beta_{52} = \beta_{61} = \beta_{63} = 0$$

$$\beta_{41} < 0, \beta_{53} < 0, \beta_{62} < 0$$



Pourquoi $\epsilon_5^M = \epsilon_6^M$ quand l'azote est au centre ?